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$$T_K = 25^\circ\text{C} + 273.15 = 298.15 \text{ K}$$

$$\Delta S^\circ = 108.7 \text{ J/K} \times \frac{1 \text{ kJ}}{1000 \text{ J}} = 0.1087 \text{ kJ/K}$$

$$\Delta H^\circ = 28.05 \text{ kJ}$$

$\Delta G^\circ = \Delta H^\circ - T \Delta S^\circ$

This state is standard state of the system

At 0°C and 101.3 kPa

At 100°C and 101.3 kPa

At 25°C and 101.3 kPa

At 20°C and 101.3 kPa

At 15°C and 101.3 kPa

At 10°C and 101.3 kPa

At 5°C and 101.3 kPa

At 0°C and 101.3 kPa

At -5°C and 101.3 kPa

At -10°C and 101.3 kPa

At -15°C and 101.3 kPa

At -20°C and 101.3 kPa

At -25°C and 101.3 kPa

At -30°C and 101.3 kPa

At -35°C and 101.3 kPa

At -40°C and 101.3 kPa

At -45°C and 101.3 kPa

At -50°C and 101.3 kPa

At -55°C and 101.3 kPa

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At -65°C and 101.3 kPa

At -70°C and 101.3 kPa

At -75°C and 101.3 kPa

At -80°C and 101.3 kPa

At -85°C and 101.3 kPa

At -90°C and 101.3 kPa

At -95°C and 101.3 kPa

At -100°C and 101.3 kPa

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